# Logarithmic Functions

Finite Math

14 February 2019



1/12

### Quiz

In the formula for exponential growth,  $A = ce^{rt}$ , if the growth rate were 5%, which letter would you replace with .05?



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Since  $(1)^2 = 1$  and  $(-1)^2 = 1$ , we get *two* values when we run  $x^2$  backward! So  $x^2$  is not invertible.

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If we have a one-to-one function

$$y = f(x)$$

we can form the *inverse function* by switching *x* and *y* and solving for *y*:

$$x = f(y) \stackrel{\text{solve for } y}{\longrightarrow} y = f^{-1}(x).$$

We will focus on one particular inverse function: the inverse of the function  $f(x) = b^x$   $(b > 0, b \ne 1)$ .

#### Definition (Logarithm)

The logarithm of base b is defined as the inverse of b<sup>x</sup>. That is,

$$y = b^x \iff x = \log_b y$$
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function	domain	range
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5/12

**Finite Math** Logarithmic Functions

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# Graphing a Logarithmic Function

#### Example

Sketch the graph of  $f(x) = \log_2 x$ .



6/12

### Property (Properties of Logarithms)

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- log<sub>b</sub> MN

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$$O \log_b M^p$$



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8 / 12

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$$\log_b x = \frac{\ln x}{\ln b}$$

(See the textbook for a proof of this.)

8 / 12

# Using Properties of Exponents and Logarithms

#### Example

Solve for *x* in the following equations:

- (a)  $7 = 2e^{0.2x}$
- (b)  $16 = 5^{3x}$
- (c)  $8000 = (x-4)^3$



# Reminder of Some Exponent Types

A quick reminder of different types of exponents:

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Finite Math Logarithmic Functions 14 February 2019 10 / 12

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•  $a^{\frac{m}{n}}$ 

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$$\bullet \ a^{\frac{m}{n}}=(a^m)^{\frac{1}{n}}$$



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$$a^{\frac{m}{n}} = (a^m)^{\frac{1}{n}} = \sqrt[n]{a^m}$$

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$$\bullet \ a^{\frac{m}{n}}=(a^m)^{\frac{1}{n}}=\sqrt[n]{a^m}=\left(\sqrt[n]{a}\right)^m$$

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### Now You Try It!

#### Example

Solve for x in the following equations:

- (a)  $75 = 25e^{-x}$
- (b)  $42 = 7^{2x+3}$
- (c)  $200 = (2x 1)^5$

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#### Solution

- (a)  $x \approx -1.09861$
- (b)  $x \approx -0.53961$
- (c)  $x \approx 1.94270$



### **Applications**

Recall that exponential growth/decay models are of the form

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.

Using the natural logarithm, we can solve for the rate of growth/decay, r, and the time elapsed, t. Let's see this in an example.

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#### Example

The isotope carbon-14 has a half-life (the time it takes for the isotope to decay to half of its original mass) of 5730 years.

- (a) At what rate does carbon-14 decay?
- (b) How long would it take for 90% of a chunk of carbon-14 to decay?

Finite Math Logarithmic Functions 14 February 2019 12 / 12